

Chemical Classes of Concern:

I. Highly Fluorinated Chemicals

What Are They and How Are They Used?

The perfluorinated and polyfluorinated chemicals are a class of highly fluorinated substances referred to as fluorochemicals for short. They include chemicals that contain one or more carbon atoms whose carbon-hydrogen bonds are replaced by carbon-fluorine bonds. Carbon-fluorine bonds are the shortest and strongest chemical bonds. When multiple numbers (abbreviated as “n”) of fully-fluorinated carbons are linked together they are known as a “C_n” perfluorinated group. Fluorochemicals have many different structures and the perfluorinated group gives them unique properties. They are used in a variety of consumer products, including cookware, clothing, carpeting, and food packaging materials, to provide an oil or water resistant finish.^{1,2} Fluorochemicals are also used as surfactants in products such as firefighting foams that are used to extinguish fuel-based fires at airports, military bases, and refineries.³

Which Fluorochemicals Are Of Concern?

Two C₈ fluorochemicals, perfluorooctanoic acid (PFOA) and perfluorooctane sulfonate (PFOS), are well studied and are associated with a range of adverse health effects. In response to concerns about these two compounds, in 2009 the U.S. EPA issued Provisional Health Advisories recommending maximum safe exposure concentrations in drinking water for PFOS and PFOA.⁴ PFOS is also a restricted chemical under the Stockholm Convention and the European Union’s REACH chemical legislation.

In the U.S. and Europe, production of PFOS, PFOA, and other molecules containing a C₈-perfluorinated group has largely ceased since 2002; however, production likely continues elsewhere. Some substitutes for C₈-based products contain C₆ and C₄ perfluorinated groups;^{5,6} the health effects of these replacement compounds are not well studied.

Many fluorochemicals available in commerce are precursors to PFOS, PFOA, and analogous shorter compounds because they transform to these compounds in the environment^{7,8} and animals.⁹ They are detected in surface waters¹⁰⁻¹² and groundwater¹³ that could be used as drinking water sources and are also found in consumer products such as food packaging materials. Thus, these precursor compounds may contribute indirectly to exposure to PFOS, PFOA, and their shorter analogs and may themselves be harmful.

Why Are They A Concern?

PFOS, PFOA, and their shorter-chain analogs are very persistent: they are inert to most natural breakdown processes¹⁴ and persist in humans, biota, and the environment for decades. Fluorochemicals, and especially PFOS and PFOA, are detected in Arctic animals such as polar bears and seals several thousand miles from manufacturing sources.¹⁵ They are also detected in humans all over the world.¹⁶

Migration of fluorochemicals out of consumer products and into food¹⁷ and household dust^{18,19} is well documented. Fluorochemicals are also widely detected in food that does not come in contact with a fluorochemical-containing packaging material.²⁰ Breast milk also contains fluorochemicals and is thought to be the primary source of exposure of these compounds for most infants.²¹ Transplacental transfer of fluorochemicals occurs between mother and fetus, with transfer potential increasing for the shorter-chain fluorochemicals.²²

PFOS and other fluorochemicals with an 8-carbon perfluorinated chain bioaccumulate in humans.²³ The oleophobic and hydrophobic properties of fluorochemicals lead to their partitioning in protein-rich compartments of the body, including blood and the liver.^{24,25} Twenty-two different fluorochemicals were recently detected in the blood of fifty non-occupationally exposed Americans ranging in age and gender;²⁶ for at least half of those compounds, there is no toxicological information publicly available.

Adverse human health effects are associated with blood concentrations of PFOS and PFOA in industrially exposed populations and the general population. A study of adults living near a fluoropolymer manufacturing facility found positive associations between the highest PFOA blood levels and testicular and kidney cancers²⁷ as well as positive associations between blood PFOA concentrations and a biomarker of liver malfunction.²⁸ In the same manufacturing-exposed region, women experienced early menopause,²⁹ children demonstrated an increased susceptibility to hypothyroidism³⁰, and both boys and girls experienced reduced hormone levels and delayed puberty³¹ in relation to PFOA or PFOS concentrations in blood. PFOA concentrations in blood and total cholesterol are positively associated in both fluorochemical-manufacturing workers³² and the general U.S. population.³³ In the general population, prenatal exposure concentrations of PFOS or PFOA are negatively associated with abdominal circumference and birth length of infants,³⁴ negatively associated with concentrations of vaccine antibodies in vaccinated children,³⁵ and positively associated with obesity in women at age 20.³⁶

Health effects associated with fluorochemicals other than PFOS and PFOA are not as extensively studied. However, a few studies indicate that shorter-chain fluorochemicals also result in adverse human health effects. Shorter chain fluorochemicals, including the C₄ and C₆ perfluorinated carboxylates and the C₄ and C₆ perfluorinated sulfonates, are observed to activate a nuclear receptor protein called a peroxisome proliferator that can induce liver tumors in mouse and human cells.³⁷ The C₆ analog of PFOA, can interact with drug binding sites at lower concentrations than ibuprofen and may interfere with drug delivery.³⁸ The concentrations of the C₄ and C₆ perfluorinated sulfonates have risen in some human blood samples as PFOS and PFOA have declined,³⁹ despite reports that the C₄ and C₆ analogs lack the potential to bioaccumulate.²³

Do We Need Them?

The presence of fluorochemicals in a wide array of consumer products has fostered an expectation that many everyday products should exhibit enhanced stain resistance and oil and water repellence. For products such as high altitude mountaineering clothing, they may provide an important safety factor. However, in light of what is known about the harmful effects of fluorochemicals, the desirability of their use in consumer products such as food packaging materials, everyday apparel, carpeting, and cookware could be questioned. In products where the performance provided by fluorochemicals has come to be expected, the tendency has been to replace PFOS and PFOA with very similar substitutes. Instead of moving from one fluorochemical to the next, the challenge for manufacturers is to innovate and develop more benign alternatives through materials innovation and green chemistry.

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